Heterocycles from Ylides. Part IV.¹ Reactivity of 4-Arylmethylene- Δ^2 pyrazolin-5-ones with Stabilized Sulphur Ylides

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4-Arylmethylene-1-phenyl- Δ^2 -pyrazolin-5-ones react with carbonyl-stabilized sulphur ylides yielding spiro-[cyclopropane-1,4'- $\Delta^{2'}$ -pyrazolin]-5'-ones, identified from spectroscopic data and chemical behaviour.

SULPHUR ylides have found a wide variety of applications in organic synthesis.^{2,3} We report the preparation of spiro[cyclopropane-1,4'- $\Delta^{2'}$ -pyrazolin]-5'-one deriva-

[†] Some simple derivatives have been reported previously.⁴
¹ Part III, P. Dalla Croce and D. Pocar, preceding paper.
² (a) C. R. Johnson, Accounts Chem. Res., 1973, 6, 344; (b)
B. M. Trast, *ibid.*, 1974, 7, 85.

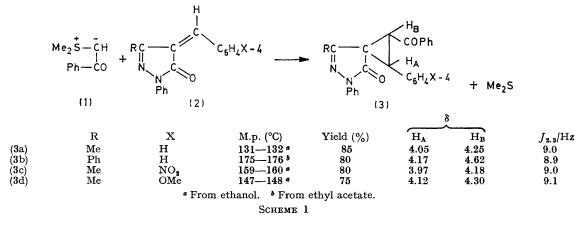
tives † from carbonyl-stabilized sulphonium ylides and 4-arylmethylene- Δ^2 -pyrazolin-5-ones.

The reaction of dimethylsulphonium phenacylide (1) with several (Z)-4-arylmethylene- Δ^2 -pyrazolin-5-ones (2)

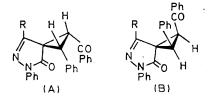
³ C. R. Johnson and C. Schroeck, J. Amer. Chem. Soc., 1973, **95**, 7417. ⁴ H. Wamhoff and F. Korte, *Chem. Ber.*, 1966, **99**, 2962.

afforded the spirocyclopropane derivatives (3a-d). Elemental analysis and i.r. data (characteristic cyclopropane bands⁵ at 1 015-1 020 cm⁻¹ and pyrazolinone $\hat{C}=\hat{O}$ stretching ⁶ at 1 680 cm⁻¹) confirmed the product structures. The n.m.r. spectra showed an AB splitting pattern for the cyclopropane hydrogen atoms, allowing us to assign to the H_AH_B system the *cis*-configuration

The structure of the spiro[cyclopropanepyrazolin]ones was supported by the following chemical transformation. Treatment of compound (3b) with potassium t-butoxide in dimethyl sulphoxide promoted abstraction of the 3proton followed by cleavage of the cyclopropane 1,2bond. This led to $4-(\alpha-benzylidenephenacyl)-1,3-di$ phenyl- Δ^2 -pyrazolin-5-one (4), which by reaction with

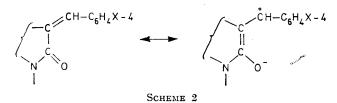


on the basis of the coupling constants 7 (J ca. 9 Hz). The products (3a-d) could therefore have structure (A) or (B).



Molecular models of compounds (3a and b) indicate that in structure (B) there exists a considerable steric interaction between the pyrazoline 3-substituent and the cyclopropane 2- and 3-substituents, whereas structure (A) appears markedly less crowded.

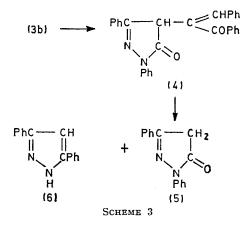
It is known⁸ that 4-arylmethylene- Δ^2 -pyrazolin-5ones show the mesomerism illustrated in Scheme 2. Accordingly, products from 1,2- or 1,4-cycloaddition



might be expected. However, the reaction gave only 1,2-cycloaddition products. This behaviour is similar to that of stabilized sulphur ylides with $\alpha\beta$ -unsaturated carbonyl compounds, affording, via Michael-type addition, cyclopropane derivatives.⁹

84, 2149.
R. H. Wiley and P. Wiley, 'Pyrazolones and Derivatives,'

hydrazine in ethanol afforded 1,3-diphenyl- Δ^2 -pyrazolin-5-one (5) and 3,5-diphenylpyrazole (6).



The mildness and simplicity of our reaction contribute to the general utility of carbonyl-stabilized sulphur ylides in this regiospecific and stereoselective route to spiro[cyclopropanepyrazolin]ones.

EXPERIMENTAL

N.m.r. spectra were measured with a Varian A-60 spectrometer (solvent C_6D_6 ; Me_4Si internal standard). I.r. spectra were recorded with a Perkin-Elmer 137 Infracord spectrophotometer for ca. 2% solutions in chloroform.

⁵ J. D. Graham and M. T. Rogers, J. Amer. Chem. Soc., 1962,

⁷ H. Booth, *Progr. NMR Spectroscopy*, 1969, **5**, 167; L. M. Jackman and S. Sternhell, 'Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry,' Pergamon, Oxford, 1969.

⁸ G. Cellerino, G. Desimoni, P. P. Righetti, and G. Tacconi,

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Elemental analyses for new compounds are available as Supplementary Publication No. SUP 21652 (2 pp.).*

4-Arylmethylene-1-phenyl- Δ^2 -pyrazolin-5-ones. These were prepared by reported methods ¹⁰ from 1-phenyl-3-methyl- or 1,3-diphenyl- Δ^2 -pyrazolin-5-ones and the appropriate aldehyde.

Preparation of Spiro[cyclopropanepyrazolin]ones (3a-d): General Procedure.—A solution of dimethylsulphonium phenacylide 9 (1) (0.01 mol) in 1,2-dimethoxyethane (25 ml) was added to a stirred suspension, cooled to -20 °C, of the 4-arylmethylene-1-phenyl- Δ^2 -pyrazolin-5-one (0.01 mol) in the same solvent (50 ml). The mixture was stirred for 2 h and then the solvent was evaporated off and the residue was crystallized.

4- $(\alpha$ -Benzylidenephenacyl)-1,3-diphenyl- Δ^2 -pyrazolin-5-one (4).—A mixture of compound (3b) (1 g) and potassium tbutoxide (0.295 g) in dimethyl sulphoxide (30 ml) was kept overnight at room temperature. The solvent was evaporated off under vacuum and the residue taken up with water

* For details of Supplementary Publications see Notice to Authors No. 7, J.C.S. Perkin I, 1975, Index issue. (25 ml) and ether (50 ml). The aqueous layer was acidified with acetic acid and the precipitate filtered off. The *product* was purified from ethanol; yield 0.75 g (75%); m.p. 130° .

Reaction of the Pyrazolinone (4) with Hydrazine.—A solution of compound (4) (1 g) in ethanol (50 ml) was treated with hydrazine hydrate (0.14 g) and refluxed for 4 h. Evaporation left an oily residue which was dissolved in ether. Extraction with aqueous 10% sodium hydroxide (20 ml) followed by acidification with hydrochloric acid (2N; 20 ml) afforded 1,3-diphenyl- Δ^2 -pyrazolin-5-one (5) (0.37 g, 70%), m.p. 134°. From the ether layer by conventional work-up, 3,5-diphenylpyrazole (6) was isolated (0.32 g, 65%); m.p. 198°. Both compounds (5) and (6) showed no m.p. depression in admixture with authentic samples.

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¹⁰ G. Desimoni, A. Gamba, P. P. Righetti, and G. Tacconi, *Gazzetta*, 1972, **102**, 491.